

GOLIK, A.Z. [Holyk, O.Z.]; SHIMANSKIY, Yu.I. [Shymans'kyi, Iu.I.]; KOBIYCHUK, N.M.
[Kobiiichuk, N.M.]

Compressibility of isoviscous substances [with summary in English].
Ukr.fiz.zhur. 3 no.4:537-541 J1-Ag '58. (MIRA 11:12)

1. Kiyevskiy gosudarstvennyy universitet.
(Compressibility)

GOLIK, A.Z.; MOCHARNYUK, R.F.

Physical properties and structure of normal alcohol solutions
in acetone. Ukr. khim. zhur. 24 no.1:29-32 '58. (MIRA 11:4)

1.Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.
(Alcohols) (Acetone) (Solution (Chemistry))

KOTORLENKO, L.A.; GOLIK, A.Z.; KOVNERISTAYA, A.S.

Viscosity and electric conductivity of lithium chloride solutions in
alcohols. Ukr.khim.zhur. 24 no.5:618-625 '58. (MIRA 12:1)

1. Kiyevskiy gosudarstvennyy universitet imeni T.G. Shevchenko.
(Lithium chloride) (Solution (Chemistry))

GOLIK, A.Z.; SOLOMKO, V.P.

Investigation of the physical properties of the water - acetone - alcohol system. Part 1: Water - acetone-ethanol system. Ukr.khim.zhur. 24 no.6:734-740 '58. (MIRA 12:3)

1. Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.
(Acetone) (Ethyl alcohol) (Systems (Chemistry))

GOLIK, A.Z. [Holyk, O.Z.]

Viscosity and electrical conductivity of zinc and cadmium amalgams. Part 2. Ukr.fiz.zhur. 4 no.4:491-496 Jl-Ag '59.
(MIRA 13:4)

1. Kiyevskiy gosudarstvennyy universitet im. P.O. Shevchenko.
(Amalgams) (Zinc) (Cadmium)

GOLIK, A.Z. [Holyk, O.Z.]; SHIMANSKAYA, Ye.T. [Shymanskaya, O.T.]

Investigation of the critical state of substances by Isoplets
method. Part 2. Temperature dependence of the density of hexane
near the critical point. Ukr.fiz.zhur. 4 no.6:769-788 M-D '59,

1. Kiyevskiy gosudarstvennyy universitet im. I.G.Shevchenko.
(Hexane--Thermal properties)

GOLIK, A.Z.; SOLOMKO, V.P.

Investigation of the physical properties of the system water-acetone-alcohols. Part 2: System water-acetone-butanol. Ukr. khim. zhur. 25 no.1:40-44 '59.

(MIRA 12:4)

1. Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.
(Water) (Acetone) (Butyl alcohol)

38

PHOTO BY DIALEKTRIC CO/5469

Sevachchiniye po kriticheskim yavleniyam i flyuktuatsiyam v
rastvorakh. Moscow, 1960.

Kriticheskkiye yavleniya i flyuktuatsii v rastvorakh; trudy
II (1960) yanvar' 1960 g. (Critical Phenomena and Fluc-
tuations in Solution; Transactions of the Conference,
January 1960) Moscow, Izd-vo Akad. Nauk, 1960. 190 p. 2,500
kopii; printed.

Sponsoring Agencies: Akademiya nauk SSSR. Otdeleniye khimi-
cheskikh nauk. Moskovskiy gosudarstvennyy universitet im.
M. V. Lomonosova. Khimicheskiy fakul'tet.

Responsible Ed.: M. I. Shchegolev, Doctor of Chemical
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Tech. Ed.: S. G. Tikhomirova.

PURPOSE : This collection of articles is intended for scientific
personnel concerned with chemistry, physics, and heat power
engineering.

Card 1/9

• Critical Phenomena and Fluctuations

SCV/B-63

COVERAGE. The book contains all the 23 reports given at the Conference on Critical Phenomena and Fluctuations in Substances organized by the Chemical Division of Soviet Academy of Sciences, January 26-28, 1980. The papers contain results of investigations carried out in recent years by Soviet physicists, statisticians, and mathematicians. The Conference, organized by the Institute of Mathematics of the USSR Academy of Sciences, was convened by Professor Kh. I. Alichanov, A. A. Belik, I. R. Krichevskiy (Chairman), V. K. Sazonovskiy, A. V. Steponkin, I. Z. Visher, and M. I. Shcheparenov (Deputy Chairman). References accompany individual articles.

TABLE OF CONTENTS:

Foreword

3

Alichanov, Kh. I., A. M. Kerimov, and B. G. Alibekov [Institut molekulyarnoy fiziki, Dagestanskiy filial AS SSSR -- Laboratory of Molecular Physics, Dagestan Branch, AS USSR]. Thermophysical Properties of Matter at Critical Temperature 5

Card 2/9

50

Critical Fluctuations and Fluctuations 36V/5460

Kazmina, L. P., and N. V. Chukhrov [Laboratory of the
Theoretical Chemistry of Solutions, Chemistry Division, Moscow
State University under I. V. L'vov]. Dielectric Properties
of Solvents in Nitro-amine -- Cyclohexane and Ethyl Alcohol -
Diethylamine Solutions

32

Khokhlov, R. M., and N. I. Chukhrov [Laboratory of the
Theoretical Chemistry of Solutions, Chemistry Division, Moscow
State University under I. V. L'vov]. Dielectric Properties
of Solvents in Electrodynamic Fields of the Millimetric
Region and Concentration Fluctuations

37

Krichevskiy, I. R., and N. Ye. Kuz'mova [Laboratory of high-pressure
divisions CIAP -- Laboratory of High-Pressure [Studies], Moscow
State Design and Planning Scientific Research Institute of
the Nitrogen Industry]. Diffusion of Liquid and Gaseous Solu-
tions in the Critical Region

45

Krichevskiy, I. R., and Yu. V. Tsekhan'skaya [Laboratory of
Card 4/9

Section IV: Summary and Conclusions

Sov/5405

Card 5/9

307/3209

Critical Phenomena and Fluctuations	
in the Critical Region, Institute of Physics, Minsk, Belarus State University.	
Effect of Critical Phenomena on the Properties and Structure of the Critical Region	81
Khlystov, N. V., and I. V. Klyuch [Institute of Physics, Minsk, Belarus State University, Institute of Physics, Minsk, Belarus State University] Variation in the Structure of the Critical Region in Relationship to the Structure of the Concentration in the Critical Region. 89	
Khlystov, N. V., and I. V. Klyuch [Institute of Physics, Minsk, Belarus State University, Institute of Physics, Minsk, Belarus State University] Variation in the Structure of the Critical Region in Relationship to the Structure of the Concentration in the Critical Region. 93	
Khlystov, N. V., D. I. Mel'nyuk, and N. G. Svirskiy [Minsk, Belarus Oblast'nyy polesnoy i lesotekhnicheskiy institut -- Forest and Technical Institute of the Minsk Oblast]. Mjesečne Izmenenija v Strukture Organicheskikh Lichidov na Konstantnoj Densnosti v Svojstvakh na Kriticheskogo Sostoyaniya	102

Card 6/9

38

Critical Phenomena and Fluctuations. SOV/5469

Re: Shcheglov, G. P. [Laboratoriya molekulyarnoy fiziki, Fiziko-
chimicheskii in-t, Kiyev, Kiyevskii gosudarstvennyi univ. s/it im.
D. I. Mendeleva -- Laboratory of Molecular Physics, Divi-
sion of Physics, Kiev State University imeni T. G. Shevchenko]
Investigation of Fluctuations in Solutions by the Method of
Light Scattering

109

Shcheglov, V. P. [Laboratoriya molekulyarnoy fiziki, Uralskiy
neftekhimicheskiy in-t im. S. M. Kirova -- Laboratory of
Molecular Physics, Ural Polytechnic Institute imeni S. M.
Kirov]. Special Structural Features of Matter in the Vicin-
ity of the Critical Point and Transfer Phenomena

117

Saripov, V. P., and Yu. D. Kolpashov [Laboratory of Molecular
Physics, Ural Polytechnic Institute imeni S. M. Kirov, and
the Laboratoriya toplofiziki, Ural'skiy filial AN SSSR --
Thermophysics Laboratory, Ural Branch, AS USSR]. Light Scat-
tering in Carbon Dioxide along Pre- and Post-Critical Isotherms

125

Smirnov, B. A. [Institut neftekhimicheskogo sinteza AN SSSR --
Card 7/9

Critical Phenomena and Fluctuations	SCV/5469
Institute of Petrochemical Synthesis, AS USSR (Moscow)] Visual Observations in the Critical Region	137
Fischer, I. Z., and V. K. Prokhorenko. Concerning the Fluctuations of Coordination Numbers in Liquids	142
Fischer, I. Z. [Belorussiy Gosudarstvennyy Universitet -- Belorussian State University (Minsk)] Correlation Analysis of the Critical Point	148
Sklyorinoy, N. I. [Laboratory of the Molecular Chemistry of Solutions, Chemistry Division, Moscow State University imeni M. V. Lomonosov]. Fluctuations in Solutions	151
Shimanskaya, Ye. T., and A. Z. Golik [Laboratory of Molecular Physics, Physics Division, Kiev State University imeni T. G. Shevchenko]. Investigation of the Critical State, Liquid-Vapor, of Solutions by Teppler's Method	161

Card 8/9

36

Critical Phenomena and Fluctuations 10/10/69

Chernomyrdin, Yu. T., Yu. I. Chernyshov, and A. S. Galin (Institute of Molecular Physics, Division of Physics, Institute of Physics and Mathematics of St. Petersburg University imeni P. C. Shorokhova). Investigations of the Critical State of Pure Substances by Taylor's Method 172

Proceedings of the Conference on Critical Phenomena and Fluctuations in Solutions 139

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10-10-01

Card 9/9

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25571
S/185/60/005/002/012/022
D274/D304

AUTHORS: Golyk, O.Z. and Cholpan, P.P.

TITLE: Molecular structure, compressibility, surface tension and viscosity of certain polysiloxanes

PERIODICAL: Ukrayins'kyy fizychnyy zhurnal, v. 5, no. 2, 1960,
242-250

TEXT: Polymethyl- and polyethylsiloxanes with linear molecules are experimentally studied, this article being a continuation of one of the authors previous works: O.Z. Golyk (Ref. 2: UkrZh, 23, no. 2, 139, 1957, and 2 articles in collaboration with others). From intensity curves of X-ray scattering, electron-density curves were constructed; these were used for determining the valence angles, the length of the chemical bond, and the packing of the molecules in the liquid state. The intensity curves, plotted on figures, show that polymethyl- and polyethylsiloxanes with linear molecules have a similar structure in the liquid state. The density, surface tension, compressibility and viscosity of these substances were

Card 1/3

Molecular structure...

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D274/D304

investigated for a wide temperature range; figures and tables are given with the results of these investigations. For polymethylsiloxanes, the polytherms of surface tension and of viscosity are the higher, and those of compressibility - the lower, the higher the potential of intermolecular forces, and the higher the critical temperature of the substance. The surface tension is also in direct proportion with the size of the molecules. Adiabatic compressibility of polymethylsiloxanes was studied by means of an ultrasonic interferometer. The temperature dependence of viscosity follows an exponential law. The polytherms of surface tension and of viscosity in the case of polyethylsiloxanes, are also the higher, the higher the potential of intermolecular forces and the higher the critical temperature. The activation energy too, is in direct proportion with intermolecular potential and critical temperature. The viscosity of binary solutions of polymethylsiloxanes was also studied, and isoviscous substances were obtained; both the activation energy and also compressibility of the isoviscous substances is practically the same. This study gives additional proof of the correspondence between structure and intermolecular forces on the

Card 2/3

Molecular structure...

25577
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D274/D304

one hand, and surface tension, compressibility, and viscosity on the other. There are 9 figures, 4 tables and 3 references: 2 Soviet-bloc and 1 non-Soviet-bloc. The reference to the English-language publication reads as follows: H.S. Green, The molecular theory of fluids, Amsterdam, 1952; I.J. Kirkwood a. F.P. Buff, J. Chem. Phys., 17, 338, 1949; I.J. Kirkwood, F. P. Buff, H.S. Green, J. Chem. Phys., 17, 998, 1949.

ASSOCIATION: Kyyvs'kogo ordena Lenina universytetu im. T.G. Shevchenka (Kiyev Order of Lenin University im. T.G. Shevchenko), Department of Molecular Physics

SUBMITTED: October 1, 1959

✓

Card 3/3

GOLIK, A.Z. [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of some poly-siloxanes. Part 2: Structure and physical properties of isoviscous polysiloxanes. Ukr. fiz. zhur. 5 no.6:843-849 N-D '60.
(MIRA 14:3)

1. Kiyevskiy ordena Lenina gosudarstvennyy universitet im. T.G. Shevchenko.

(Siloxanes)

GOLIK, A.Z. [Holyk, O.Z.]; CHOLPAH, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of some polysiloxanes.
Part 3: Viscosity, compressibility, and structure of liquid cyclic
polysiloxanes. Ukr. fiz. zhur. 5 no.6:850-856 N-D '60.
(MIRA 14:3)

1. Kiyevskiy ordena Lenina gosudarstvennyy universitet im. T. G.
Shevchenko.

(Siloxanes)

SKRYSHEVSKIY, Anton Frantsovich; COLIK, A.Z., prof., ctv. red.;
DROZHIN, E.V., red.; OKOITAYA, Ye.D., tekhn. red.

[Diffraction of X rays, electrons, and neutrons in gases and
the molecular structure] Difraktsiya rentgenovskikh luchei,
elektronov i neitronov v gazakh i stroyenie : elekul. Kiev, Izd-
vo Kievskogo univ., 1961. 84 p. (MIRA 15:9)
(X rays--Diffraction) (Electron diffraction examination)
(Neutrons--Diffraction)

GOLIK, A.Z.; CHOLIK, P.F.

Speed of ultrasound in some polysiloxanes. Akust. zhur. 7 no.1:33-39
'61. (MKA 14:4)

1. Kiyevskiy gosudarstvenny universitet.
(Siloxanes)
(Ultrasonic waves)

5/08 4/000/021/0-0/094
E-C2/2158

AUTHORS: Shimanskaya Ye. T., Shimanskaya I. I., Golik, A. S.

TITLE: Investigation of the critical state of pure substances by
Tepier's method

PERIODICAL: Referativnyy zhurnal. Khimika no. 21 '96, 43, abstract
21B347 (Sb. "Kritich. yavleniya i flyuktuatsii v rasvorakh",
M., AN SSSR, '960 17' '88)

TEXT: A method has been developed, for the investigation of critical
states, by means of which the density ρ of a substance can be measured
in any point in a chamber (by the optical Tepier method) with long-time
thermostating. The apparatus is described in detail. Heptane and hexane
were examined. Density has a non-monotonic gradient with respect to the
chamber height Z and has a maximum at the meniscus. This maximum
increases as the temperature approaches the point T_m at which the meniscus
vanishes. With a steady temperature change rate, $\sim 1^\circ\text{C}/\text{hr}$, the $d\rho/dZ$
maximum is present on heating and absent on cooling (i.e., a hysteresis
is observed). With irregular changes in temperature and long-time
Card 1/

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E102/B138

Investigation of the critical

At 20 hr^{-1} thermostating, the $d\eta/dt$ maximum is however present on heating as well as cooling; the maxima are then lower than in the case of steady heating. The $\eta(t)$ curves are found by integrating $d\eta/dt = f(t)$. For $T \sim T_m$ they are S-shaped. In the lower part of the chamber density is higher, and in the upper part lower than critical. This is in full agreement with classical representations regarding the existence of a critical point and not a region when allowing for the effect of gravitational field. The critical state is realized only in a narrow layer at the point where the mentioned vanishes. Above and below this layer the substance is not in a critical state although its temperature is critical. As the density difference throughout the chamber corresponds at the critical temperature to the equilibrium state, then it must be assumed that displacement sometimes occurs, levelling the density and removing the system from the state of equilibrium. [Abstracter's note: Complete translation.]

Card 2/2

GOLIK, A.E.; BAKHOVSKIY, V.Ye.

Heat of vaporization, composition of vapors, and surface tension of solutions of paraffins and alcohols. Ukr.khim.zhur. 27 no.5: 574-577 '61. (MIFI 14:9)

1. Klyovskiy pedagogicheskiy universitet im. T.G. Shevchenko.
(Paraffins) (Alcohols)

YOLIK, A.Z.; BAKHOVSKIY, V.Ye.

Latent heat of vaporization of alcohols in acetone solutions.
Ukrainian. 27 no.5:577-580 '61. (MIR. 14:9)

1. Kiyevskiy gosudarstvennyy universitet im. T.G. Shevchenko.
(Alcohols) (Heat of vaporization)

15 8:10

67
S/073/617/017/06/01/00
8110/8117

AUTHORS: Golik, A. Z., Cholpan, P. F., Ivanchuk, I. I.

TITLE: Investigation of some physical properties of some
phenyl siloxanesPERIODICAL: Ukrainskiy khimicheskiy zhurnal, v. 17, no. 6, 1973
754 - 759

TEXT. This work is an investigation of viscosity, supercooled specific and
adiabatic compressibility of 1,5-dimethylphenyl-3-methylphenyl-2,
siloxane $(CH_3)_2C_6H_5SiOSi(C_6H_5CH_3)OSiC_6H_5(CH_3)_2$; 1,5-dimethyl-1
methylphenyltrisiloxane $(CH_3)_2SiOSi(C_6H_5CH_3)OSi(CH_3)_2$; 1,5-dimethyl-1
methylphenyltetrasiloxane $(CH_3)_3Si[OSiCH_3C_6H_5]_2OSi(CH_3)_3$; polymer 2 (P2) $(CH_3)_3Si[OSiCH_3C_6H_5]_2$
 $1 (P1) (CH_3)_3Si[OSiCH_3C_6H_5]_2OSi(CH_3)_3$; polymer 2 (P2) $(CH_3)_3Si[OSi(CH_3)_2]_2$
 $[OSi(CH_3)_2]_2; OSi(CH_3)_3$; polymer 3 (P3) $(CH_3)_3Si[OSi(CH_3)_2]_2$
 $[OSi(CH_3)_2]_2$.

Card 1/6

8

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B.I. 6/11"

Investigation of some physical...

-OSi(CH₃)₃; polyene 1 (P1) (CH₂)₂Si[OSi(CH₃)₃]₂OSi(CH₃)₃

structure of the polymethyl phenyl siloxane was studied and its structure and their molecular weight was determined. The viscosity of the polymer with temperature according to the viscometric method was measured. The polytherms of the viscosities of P₁ and P2 and the polyether of the siloxane. Only the first three substances correspond to the following structures:

$\eta_{sp} C/(V + d)$ (Table 1) Between 0° and 100°C, for the polymer of the siloxane in rearing temperature. Ultrasonic speed was measured at 25°C by the interferometer by I. G. Mikhailev (1967). The speed of the wave was calculated according to the formula $\lambda = C/V$, where C is the speed of the length of ultrasonic wave. In general, for the polymer of the siloxane, the trimer with 4 C₂H₅ groups (#), of the trimers with C₂H₅ and of the methyl trimer (C), and of the tetramer (D), the speed of the tetramer with two C₂H₅ groups decreases little in the range of temperature, with the polytherm of A (viscometric) and the polytherm of the

Page 2/3

Investigation of some physical...

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B110/B147

polytherm of D lies above that of E. In P1, P2, and P3, a slight deviation from linearity was found at 40°C (near their solidification point). Adiabatic compressibility was calculated by; $\beta = 1/a^2 \rho$ (a = ultrasonic speed, ρ = density, β = adiabatic compressibility. It is inversely proportional to the number of phenyl radicals. From the linear dependence; $\ln \beta = f(t)$, $\beta = \beta_0 \exp(T/C)$ is derived; T = experimental temperature, β_0 = adiabatic compressibility at $T = 0$, C = constant (Table 2). There are 10 figures, 2 tables, and 3 Soviet references.

ASSOCIATION: Kiyevskiy gosudarstvennyy universitet im. T. G. Shevchenko
(Kiyev State University imeni T. G. Shevchenko)

SUBMITTED: September 29, 1960

Card 3/5

GOLIK, A.Z.; KLASSEN, I.F.; KUCHAK, G.M.

Speed of propagation of ultrasonic waves in certain zinc and cadmium amalgams. Akust. zhur. 7 no.2:258-260 '61. (MIRA 14:7)

1. Kiyevskiy gosudarstvennyy universitet.
(Ultrasonic waves--Speed) (Zinc amalgam)
(Cadmium amalgam)

GOLIK, A.Z., ref., etv. red.; ROSHCHINA, G.P., dots., etv. red.;
MIRONETS, Ye.M., red.; KHOKHANOVSKAYA, T.I., tekha. red.

[Structure and physical properties of matter in the liquid state; materials] Stroenie i fizicheskie svoistva veshchestva v zhidknom sostoyanii; materialy. Kiev, Izd-vo Kievskogo univ., 1962. 146 p. (PIRA 15:2)

1. Sovershchennye posvyashchennye problemy zhidkogo sostoyaniya veshchestva. 4th, Kiev, 1950.
(Liquid)

GOLIK A.Z.

STRUCTURE AND PHYSICAL PROPERTIES OF MATTER IN A LIQUID STATE
 Reports read at the 4th Conference convened in KIEV from 1 to 5 June
 1959, published by the publisher House of KIEV University, KIEV,
 USSR, 1962

A.S. GOLIK and I.P. AL'GEN, Connection Between Viscosity and Electrical Conductivity and the Structure of Zinc and Calcium Alkaloids	96
A.S. LASHKO, Rontgenographic Investigation of the Liquid Au-In Alloy	101
A.V. ROPANOV and I.A. KHO, Rontgenographic Investi- gation of the Structure of Tin-Lead Liquid Alloys	107
Y.A. GEMAL'OV, A.V. NIKOL'SKAYA and A.P. TSVETOV, Thermodynamic Properties of Liquid Metallic Alloys	115
V.I. GOREVSKY and D. SUDOV, Investigation into A transition layer on a liquid metallic surface	118
V.V. PAVLENKO, on the Multi-Types of phase Transitions	124
V.V. PAVLENKO and V.V. VENKOVSKII, Dielectric Parameter of the Liquid Alloys, and whether the Critical Region in the Liquid Alloys	129

GOLIK, A.Z.

STRUCTURE AND PHYSICAL PROPERTIES OF MATTER IN A LIQUID STATE
 reports read at the 4th Conference convened in KIEV from 1 to 5 June
 1969, published by the Institute of Physics of KIEV University, Kiev,
 USSR, 1969.

A.Z. GOLIK and L.A. BOLIK, Molecular Structure, Correlation, Surface Tension and Viscosity of some Polysiloxanes	57
N.N. GORODIKOV, Influence of Viscosity on Electrokinetic Properties of Liquids	65
O.YA. S. BOYMOV, Connection Between the Coordination Number and the Thermal Lattice of Aqueous Solution Particles of Electrolytes	71
I.G. TIKH YEV and YU.YA. S. BOYMOV, Thermal Dependence of the Ionotropic Impermeability of the Aqueous Solution of Ions at Low Concentrations	74
M. A. BELYI and S. V. KUDRIK, The Effect of Solvents and Salt Concentration on the Ionotropic Impermeability of the Salt Solutions	79
YU.YA. S. BOYMOV, K.Y. SUDNIKOV and V.A. KH. VYEV, Theory of Ultrasonic Absorption in Electrolyte Solutions	81
G. G. VARYASHVICH, Connection Between the Structural Units of Gases and Structural Units of Liquids	97

ROSMCHINA, Galina Petrovna; VOLIK, I.Z., prof., otd. red.; VZLIO,
V.I., red.; V. K. M. V. T. M., tel'kin red.

[M]ulticolor scattering of light in pure [M]ultichromatic par-
selni sveta v vodakh, Kiev, Izd-vo Kievskoie vyd-va, 1981,
27 p. (MIRK 1511)
(Color effect) (Scattering (Physical)) (Light--Scattering)

GOLIK, A.Z., prof., otv. red.; OSHCHINA, G.P., dots., otv. red.;
- MIRONETS, Ye.M., red.; KUCHKANOVSKAYA, T.I., tekhn.red.

[Structure and physical properties of matter in the liquid
state; materials] Stroenie i fizicheskie svoistva veshchestva
v zhidkem sostoianii; materialy. Kiev, Izd-vo Kievskogo univ.,
1962. 146 p. (MIRA 15:9)

1. Soveshchaniye posvyashchennoye probleme zhidkogo sostoyaniya
veshchestva. 4th, Kiev, 1959.

(Liquids)

"APPROVED FOR RELEASE: 09/24/2001

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APPROVED FOR RELEASE: 09/24/2001

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5/843/62/000/000/006/010
5207/5308

AUTHOR OR: Golik, A.I. and Klassen, I.F.

TITLE: Relationship of the viscosity and electrical conductivity with the structure of zinc and cadmium amalgams

NOTES: Otryazhivye i fizicheskiye svoystva veshchestva v zhidkem sostoyanii; materialy IV soveshch. po probl. zhidkogo sost. veshchestva, v Kiyev'e 1959 g. Kiev, Izd-vo Kiev. univ., 1962, 96-100

TEXT: The purpose of this work was to check the hypothesis that both the first (shear) viscosity and the electrical conductivity of liquid metals and their solutions are related to the short-range order. The viscosity, density and electrical conductivity of zinc and cadmium amalgams were measured at temperatures up to 350°C in a wide range of compositions. Amalgams with the same viscosity had the same short-range order but different electrical conductivities. Amalgams with the same electrical conductivity had practically the

Card 1/2

Relationship of the viscosity ...

3/845/62/000/000/006/010
b207/b303

same density but different viscosities. The results confirm the hypothesis cited above. There are 4 figures and 1 table.

ASSOCIATION: Kiyevskiy gosudarstvennyy universitet (Kiev State University)

Card 2/2

GOLIK, A.Z.; RYNDICH, N.A.; KUCHINKA, M.Yu.; ANDRIYENKO, S.S.

Thermomechanical properties of cord made from polycaprolactam.
Khim.volok. no.2:23-25 '62. (MIRA 15:4)

1. Kiyevskiy gosudarstvenny universitet im. Shevchenko.
(Textile fibers, Synthetic) (Azepinone)

GOLIK, A.Z., [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of certain siloxanes.
Part 4. Density of two-component solutions of liquid siloxanes.
Ukr.fiz.zhur. 7 no.5:549-553 My '62. (MIRA 16:1)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko.
(Siloxanes)

GOLIK, A.Z. [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Molecular structure and physical properties of certain siloxanes.
Part 5. Surface tension and molecular interaction of liquid
siloxanes. Ukr.fiz.zhur. 7 no.5:554-558 My '62. (MIRA 16:1)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko.
(Siloxanes)

GOLIK, A.Z. [Holyk, O.Z.]; CHOLPAN, P.F. [Cholpan, P.P.]

Density and short-range coordination of certain liquids. Ukr.
fiz. zhur. 7 no. 5:559-562 My '62. (MIRA 16:1)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko.
(Liquids)

GOLIK, A.Z.

Present stage and problems of the physics of liquids. Ukr.fiz.
zhur. 7 no.7:685-686 Jl '62. (MIRA 15:12)
(Liquids)

8/185/62/007/008/001/008
D234/D303

AUTHOR: Golik, A.Z.

TITLE: Connection of compressibility and sheer viscosity
with the structure of liquid state of matter

PERIODICAL: Ukrayins'kyy fizychnyy zhurnal, v. 7, no. 8, 1962,
806 - 811

TEXT: The author gives the experimental values of compressibilities and shear viscosities of several paraffins, methylsiloxanes and alcohols, obtained by him in collaboration with P.F. Cholpan and I.I. Ivanova. Graphs of temperature dependence of these quantities are given for all substances mentioned. The compressibility was determined from data on density and ultrasound velocity, the latter being measured by I.T. Mikhaylov's interferometer. The empirical formula for the temperature dependence of the compressibility $\beta = \beta_0 \exp [\alpha T]$ is found to agree with experiment better than the expression given by the cell theory. The temperature dependence of viscosity is described by Ya.I. Frenkel's formula $\eta = A \exp$
Card 1/2

Connection of compressibility and ... S/135/62/007/008/001/008
D254/D308

[B/RT]. A table of critical temperatures and the values of ρ_0 , α , β and A is included. It is found that there is a correlation between B and α . Properties of isoviscous substances for each group were studied. The author gives as an example the graph of intensity of scattered X-rays, plotted against the scattering angle, for $(\text{CH}_3)_{10}\text{Si}_4\text{O}_3$ and the solution consisting of 48.73% of $(\text{CH}_3)_{18}\text{Si}_3\text{O}_2$ and 51.27% of $(\text{CH}_3)_{12}\text{Si}_3\text{O}_4$, isoviscous with the former. All experimental points are situated on the same curve, which indicates that the isoviscous substances have the same structure. There is 1 table and 8 figures.

ASSOCIATION: Kiyevskiy universitet (Kiev University)

Card 2/2

Surfaces of constant curvature

Carbohydrates

BARANOVSKIY, V.Ye.; SHIMANSKIY, Yu.I.; GOLIK, A.Z.

Heat of evaporation of the ternary system ethyl alcohol-butyl
alcohol - acetone. Ukr.khim.zhur. 28 no.4:484-486 '62.

1. Kiyevskiy gosudarstvennyy universitet imeni T.G.Shevchenko.
(Ethyl alcohol) (Butyl alcohol) (Acetone)
(Heat of evaporation)

GOLIK, A.Z.; RYNDICH, N.A.; NUZHNYY, V.M.; CALAGAN, Yu.

Velocity of ultrasound and the compressibility of alcohol -
acetone - water solutions. Ukr.khim.zhur. 28 no.4:506-510 '62.

(MIRA 15:8)

1. Kiyevskiy gosudarstvennyy universitet imeni T.G.Shevchenko.
(Alcohols) (Acetone) (Ultrasonic waves—Speed)

GOLIK, A.Z.; IVANOVA, I.I.

Molecular structure, density, compressibility, and shearing viscosity of n.paraffins in the liquid state. Zhur.fiz.khim. 36 no.8:1768-1770 Ag '62. (MIRA 15:3)

1. Kiyevskiy gosudarstvennyy universitet,
(Liquids) (Paraffins)

GOLIK, A.Z. [Holyk, O.Z.]; KUCHINKA, M.Yu. [Kuchynka, M.IU]

Temperature-time dependence of the strength of polymers at a
constant tension rate. Ukr. fiz. zhur. 8 no.4:479-486 Ap '63.
(MIRA 16:8)

1. Kiyevskiy gosudarstvennyy universitet im. Shevchenko,
(Polymers—Testing)

BARANOVSKIY, V. Ye., COLIK, A. Z.

Latent heat of vaporization of water-alcohol solutions, V. Ye.
Khim. zhur. 29 no. 2:157-14. 1953. (MIRA 16:5)

1. Kiyevskiy gosudarstvennyy universitet im. T. G. Shevchenko.
(Heat of evaporation) (Alcohols)

GOLIK, A.Z.; ADAMENKO, I.I.; SHOLPAK, P.F.

Effect of molecular interaction on the compressibility and
viscosity of liquids. Ukr. fiz. zhur. 9 no.4:412-416 Ap '64.
(MIRA 17:8)
1. Kiyevskiy gosudarstvennyy universitet.

"APPROVED FOR RELEASE: 09/24/2001

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APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0"

"APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0

1. *Agave* (L.) *leucophylla* (L.) *leucophylla* (L.)

1. *What is the name of the author?* *What is the title of the book?*

1. *What is the relationship between the two main characters?*

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L 51442-65 EWT(m)/EPF(c) Pr-4 RM
ACCESSION NR: AP5011070

UR/0105/65/010/004/0443/0449

AUTHOR: Holyk, O. Z. (Golik, A. Z.); Adamenko, I. I.

TITLE: Compressibility and molecular structure of liquids. I. Compressibility of n-paraffins and of their mutual solutions

SOURCE: Ukrayins'kyj fizichnyj zhurnal, v. 10, no. 4, 1965, 443-449

TOPIC TAGS: n-paraffin, molecular structure, compressibility, liquid state, activation energy, viscous flow, intermolecular force

ABSTRACT: The authors investigate the compressibility of liquids having an identical molecular structure and the same type of intermolecular forces (the n-paraffins: n-heptane, n-octane, n-nonane, n-undecane, and n-dodecane). It is shown that the compressibility polytherms of these liquids lie the lower the deeper the potential well on the molecular interaction curve and the larger the activation energy of viscous flow. It is also shown that under certain conditions it is possible to attain coincidence of the polytherms of compressibility of solutions of paraffins and pure substances or other solutions of paraffins of different composi-

Card 1/2

L 51442-65

ACCESSION NR: AP5C11070

tions (these substances are called iso-compressible). Iso-compressible substances have identical activation energy of viscous flow and identical energy of intermolecular interaction. The adiabatic compressibility polytherms of the investigated normal paraffins are well described by the empirical formula $\beta_{ad} = \beta_0 \exp \alpha T$ in which the constant α is inversely proportional to the viscous-flow activation energy. The dependence of the adiabatic compressibility on the potential of the intermolecular interaction is in good agreement with modern statistical and model theories of liquids. Orig. art. has: 5 figures, 6 formulas, and 4 tables.

ASSOCIATION: Kyyiv's'kyj dershuniversytet im. T. G. Shevchenka [Kiyevskiy gosudarstvennyy universitet im. T. G. Shevchenko] (Kiev State University)

SUBMITTED: 21Nov64

ENCL: 00

SUB COM: 00, ME

MR REF SGV: 003

OTHER: 003

me
Card 2/2

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APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0"

ACC NR: AP7004553

SOURCE CODE: UR/0185/66/011/007/0797/0801

AUTHOR: Golik, A. Z.; Cholpan, P. P.; Tarasenko, O. V.

ORG: Kiev State University im. T.H. Shevchenko (Ukrayins'kyj derzhuniversytet)

TITLE: Velocity of ultrasonic vibrations and compressibility of liquid siloxanes

SOURCE: Ukrayins'kyj fizichnyj zhurnal, v. 11, no. 7, 1966, 797-801

TOPIC TAGS: siloxane, temperature dependence, ultrasonic vibration

ABSTRACT: The authors investigated the temperature dependence (within the range of 0 - 200°C) between the velocity of ultrasonic vibrations and the adiabatic compressibility of linear methylsiloxanes - octaethyltrisiloxane, deca-methyltetrasiloxane, dodecamethylpentasiloxane, cyclic methylsiloxanes - octamethylcyclotetrasiloxane, decamethylcyclopentasiloxane, and methylphenyl-siloxanes - heptamethylphenyltrisiloxane, pentadethyltrisiloxane, octa-methyldiphenyltetrasiloxane.

It is determined that the temperature dependence of ultrasonic velocity at high temperatures deviates from the linear dependence. The adiabatic compressibility obeys an exponential law over a small range of temperatures only. It is shown that the compressibility of siloxanes decreases with the increase of the intermolecular force potential and the co-ordination number.

Orig. art. has: 4 figures, 3 formulas and 2 tables. [JPRS: 37.330]

SUB CODE: 20,07 / SUBM DATE: 11Dec65 / ORIG REFI: 009

Card 1/1

JPRS 37.330

GOLIK, F.K., vrach

Simplified apparatus for the simultaneous injection of oxygen and solutions. Zdrav. Kazakh. 17 no.9:48-50 '57.

(MIRA 12:6)

1. Iz oblastnogo kozhno-venerologicheskogo dispensera Severo-Kazakhstanskoy oblasti Kazakhskoy SSR.

(MEDICAL INSTRUMENTS AND APPARATUS) (INJECTIONS, HYPODERMIC)

65-38-6-4/43

AUTHOR: Golik, G., Senior Inspector-Pilot, DOSAAF Republic Committee of the Ukraine (Kiev) (Respublikanskiy komitet DOSAAF Ukraine)

TITLE: Model-airplane Builders Prepared for Sports Combat (Aviamodellisti, gotovy k sportivnoy bor'be)

PERIODICAL: Vechernye novosti, 1958, No. 6, p. 2 (USSR)

ABSTRACT: The author states that teams of some 270,000 model airplane builders are now being trained by 4,000 public instructors in DOSAAF primary organizations in the Ukraine. Personalities mentioned include: USSR champion Ye. Kondratenko, N. Dem'yanchuk, Ye. Kucherov, Yu. Golik, M. Cherkasskiy, V. Sheremet. In June and July (1958), 22,000 model airplane builders will compete at the Spartacus Games.

ASSOCIATION: DOSAAF Republic Committee of the Ukraine

1. Airplane-model building

Card 1/1

GOLIK, G.

At gatherings. Kryl. rod. 14 no.2:18 F 162.
(MIRA 16:4)

1. Nachal'nik otdela aviationsionnoy podgotovki respublikanskogo
komiteata Dobrovol'nogo obshchestva sodeystviya armii,
aviatsii i flotu.

(Ukraine—Parachuting)

ACC NR: AD-00144

SOURCE CODE: UR/0079/66/636/169/1636

AUTHOR: Sh. A., V. A.; Golik, I. A.; Libman, B. Ya.; Derkach, G. I.

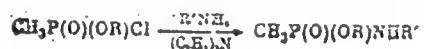
ORG: Institute of Organic Chemistry, Academy of Sciences, UkrSSR (Institut organicheskoy khimii Akademii nauk UkrSSR)

TITLE: Monoalkylamides of alkyl methylphosphonates

SOURCE: Zhurnal obshchey khimii, v. 36, no. 9, 1966, 1636-1639

TOPIC TAGS: insecticide, monoalkylaminoalkyl-methyl-phosphonate, ORGANIC AMIDE, PHOSPHONATE, PHOSPHONIC ACID

ABSTRACT: In a search for new insecticides, a series of monoalkylamides of alkyl methylphosphonates was obtained by the reaction of methylphosphonic acid chloride with primary amines in the presence of triethylamine in an ether solution at room temperature:



Composition and properties of the amides are given in the table.

Card 1/4

UDC: 547.26'118

ACC NR: AP6031382 Table 1. Monoalkylamides of alkyl methylphosphonates

$\text{CH}_3\text{P}(\text{O})(\text{OAr})\text{NHR}'$

R	R'	yield, %	bp (p, mm)	d_4^{20}	n_{D}^{20}
CH ₃	CH ₃	a, 37	72-73° (0.02)	1.1283	1.4423
CH ₃	C ₂ H ₅	a, 58	78-79 (0.02)	1.0779	1.4402
CH ₃	iso-C ₃ H ₇	a, 42	81-83 (0.03)	1.0402	1.4573
CH ₃	u.-C ₄ H ₉	a, 36	95-96 (0.1)	1.0192	1.4424
C ₂ H ₅	CH ₃	b, 82 (69)	86-88 (0.5)	1.0835	1.4550
C ₂ H ₅	C ₂ H ₅	b, 72	91-93 (0.4)	1.0482	1.4372
C ₂ H ₅	iso-C ₃ H ₇	b, 78 (62)	66-67 (0.03)	0.9935	1.4317
C ₂ H ₅	u.-C ₄ H ₉ **	a, 54 (11)	100-101 (0.1)	0.9971	1.4470
iso-C ₃ H ₇	CH ₃	b, 81 (55)	73-75 (0.06)	1.0371	1.4313
iso-C ₃ H ₇	C ₂ H ₅	b, 79	69-71 (0.03)	1.0403	1.4348
iso-C ₃ H ₇	iso-C ₃ H ₇	b, 63	83-87 (0.57)	0.9863	1.4318
Card 2/4	iso-C ₃ H ₇	a, 54 (13)	108-120 (44)	0.9712	1.4376

ACC NR: AP

NAME		PERCENT		CALCULATED	
FORMULA	NAME	PERCENT	FORMULA	NAME	PERCENT
28.89	29.13	N 41.43	C ₅ H ₁₉ NO ₂ P	N 41.55	
33.54	33.65	CH ₃ O 22.53	C ₄ H ₁₂ NO ₂ P	CH ₃ O 22.55	
38.12	38.36	CH ₃ O 20.65	C ₅ H ₁₄ NO ₂ P	CH ₃ O 20.55	
42.92	42.98	CH ₃ O 18.74	C ₆ H ₁₆ NO ₂ P	CH ₃ O 18.79	
33.32	33.65	N 10.21	C ₅ H ₁₂ NO ₂ P	N 10.22	
37.92	38.36	N 9.22; P 20.33	C ₆ H ₁₄ NO ₂ P	N 9.27; P 20.40	
43.08	42.98	P 18.59	C ₆ H ₁₆ NO ₂ P	P 18.75	
47.33	47.60	N 7.58	C ₆ H ₁₈ NO ₂ P	N 7.61	
38.03	38.36	N 9.34	C ₆ H ₁₄ NO ₂ P	N 9.27	
42.53	42.98	N 8.43	C ₆ H ₁₃ NO ₂ P	N 8.45	
47.11	47.60	N 7.99; P 17.34	C ₇ H ₁₈ NO ₂ P	N 8.01; P 17.28	
52.13	52.22	N 7.28; P 16.04	C ₈ H ₂₀ NO ₂ P	N 7.25; P 16.05	

Card 3/4

ACC NR: AP6031382

These amides have strong insecticidal properties but are very toxic to domestic animals. Monoalkylamides of alkyl methyphosphonates react with tert-butyl hypochlorite to form N-chloro-N-alkylamides of alkyl methylphosphonates. The reaction takes place in chloroform at 20-30°C. [WA-50; CBE No. 12]

SUB CODE:06,07/ SUBM DATE: 17Jul65/ ORIG REF: 003/ OTH REF: 014/

Card 4/4

KHODCHENKO, L.P., inzhener; GOLIK, G.I., inzhener.

Standard metallic edge fittings for construction yards.
Shakht.stroi. no. 4:25-27 Ap '57. (MLRA 10:7)
(Building materials industry--Equipment and supplies)

GOLIK, G.K., student IV Kursa; YEDOLETS, I.P., student V Kursa

Professor Petr Ivanovich Shatilov, founder of the original
Russian school of therapeutics. Klin.med. j. no.3:37-91
A. 1956. (MIEA 12:8)

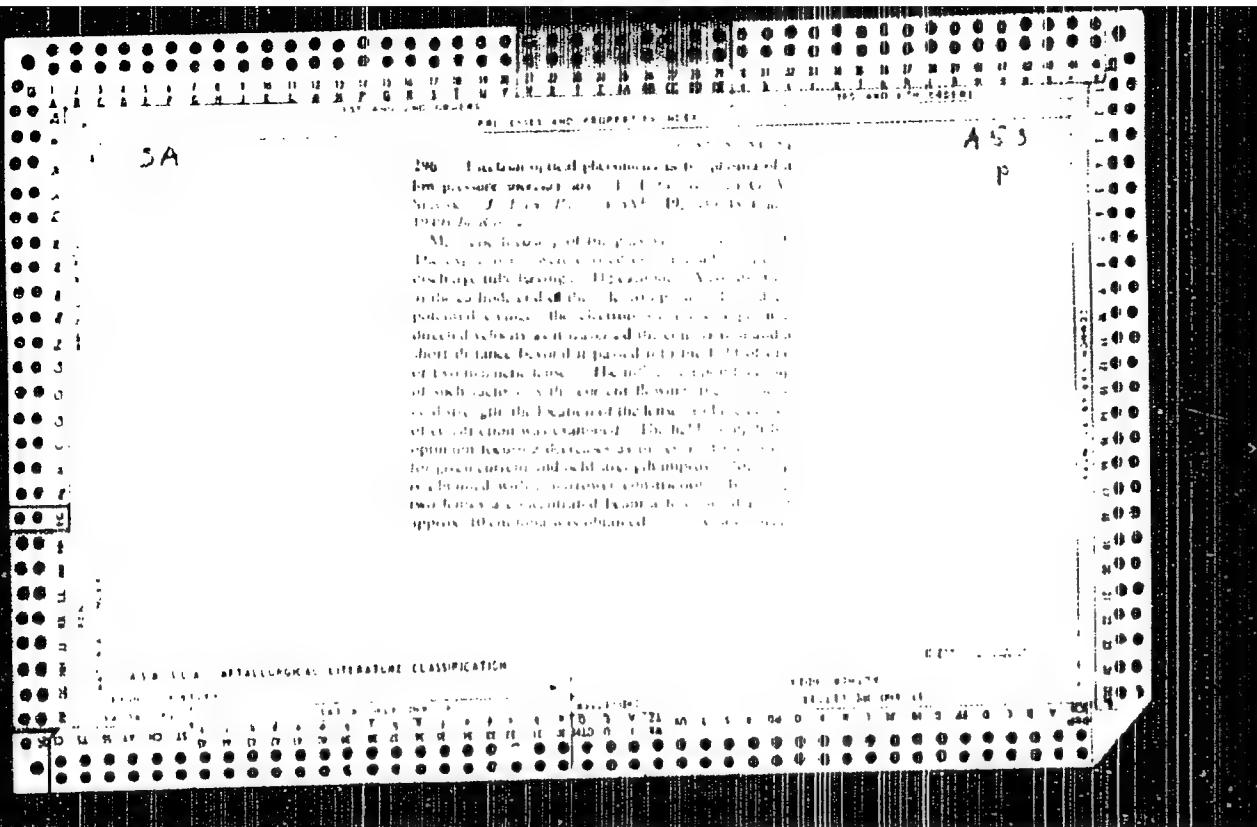
1. Iz kafedry propedevtiki vnutrennikh bolezney (zav. -
zasluzhennyj leit tel' nauki prof. V.M.Kozan- Yatenyy) lechebnogo
Fakulteta Khar'kovskogo meditsinskogo instituta i 26-v klini-
cheskoy bol'ницы (glavnnyj vrach M.M.Gorodnichenko).

(310GBuPHIES
Shatilov, Petr I.)

GOLIK, A.M. (and family) and M.V. (M. Barabash) (Novosibirsk)

Large quantity of machine and electronic components, including computer parts,
45 Novosibirsk, S-163. (CIA-1010)

1. Machine parts and small electronic components (for GOLIK).
(CPU, RAM and Freightage)



"APPROVED FOR RELEASE: 09/24/2001

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ALL INFORMATION CONTAINED

APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0"

VLIK, K.N. [Holyk, K.N.]

Diurnal and seasonal dynamics of the intensity of photosynthesis in
the sweet cherry, cherry, plum and apricot. Ukr. bot. zhur. 19 no.3:
20-27 '6. (MIRA 15:7)

1. Institut botaniki AN UkrSSR, otdel fotosintesa
(Photosynthesis) (Fruit trees)

GOMIK, ' ,Ns [holz] , - 11.]

Openness of photosynthesis and transpiration in the leaves
of different kinds of the green in the sweet cherry, sour
cherry, plum and maraschino cherries. (Licht. Wär. test.
of the leaves. 3-40. 1954)

1. Theoretical physiological picture. V. G. Gomik, sidel. fotosyn-
thesis.

GOLIK, K.N. [Holyk, K.N.]

Effect of shade on the intensity of photosynthesis and
transpiration in the representatives of Prunoideae. Ukr. bot.
zhur. v. 18, 18-24. 1964. (CISRA 17:1)

1. Otdel fotosinteza Instituta botaniki AN UkrSSR.

"APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0

CONFIDENTIAL
REF ID: A2

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APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0"

the following table gives the results of the experiments made on the growth of the *Leucosphaera* and *Leucosphaera* *var.* *lutea* and the formation of the spores of *Leucosphaera* *var.* *lutea* on the basis of the experiments of Kosteletzky (1926, 1934, 1941, 1942). The results of the possible removal of the seed of *Leucosphaera* *var.* *lutea* are given in the last column.

APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0"

Jul 49

USSR/Physics
Electron Microscope

Arce, Mercury

"Electronic Optical Phenomena in the Plasma of a Mercury Arc Under Low Pressure," I. I. Golik, S. V. Sptval, Phys. Faculty, Moscow State University M. V. Lomonosov, 9 pp

"Zhur. Tekh. Fiz." Vol. XIX, No 7

Shows that in the field of a magnetic lens acting on a stationary plasma, contractions of two types develop: (1) diffusive-plasmatic, and (2) electronic optic. Second case develops for a rapid change in potential and an artificial contraction in plasma. Measured distribution of charges along radius of ray and clarified influence of second magnetic lens capturing focusing, magnitude of discharge current influences focusing, as it does all effects of the magnetic field in the plasma. Decrease in contraction with increase in current is due to increase in proportion of speed of chaotic motion of the electrons in comparison with direction. Submitted 29 Jun 42.

51/49559

GOLIK, I. I.

USSR/Physics - Plasma Oct 53

"Electron Optical Phenomena in Focussing and
Stationary Plasma in Mercury Vapors," I.I. Golik
(deceased) and G.V. Spivak, Chair of Electron Optics
Vest Mch Univ, Ser Fizikomat i Yest Mchuk, No 7,
pp 117-123

In their previous works (see Zhur Eksp Teor Fiz, Vol 23, 1953) at their Lab the authors established the presence of the phenomena of plasma focussing, which occurs under the action of external and internal electrical and magnetic fields. Their purpose here is to study the phenomena of convection

27397

(necking) of stationary plasma toward the axis of symmetry in a strong and concentrated external magnetic field.

YALIK, L. Z., VOLKOVSKY, I. I., and SITENKEL'BERG, N. I.

"Application of a Differential Thermo-Graph for the
Investigation of Metal Transitions in Drying Silicate
Materials."

Report submitted for the Conference on Heat and Metal Transitions,
Minsk, USSR, June 1961.

卷之三

19. *Leucosia* *leucostoma* *leucostoma* *leucostoma* *leucostoma* *leucostoma* *leucostoma*

26.2421

Authors: G. W. H. Evans, T. S., and H. C. W.

1. *What is the primary purpose of the study?* (check all that apply)

PERIODICALS RECEIVED IN THE LIBRARY, NOVEMBER, 1917

APPROVED FOR RELEASE: 09/24/2001

CIA-RDP86-00513R000515720010-0"

132

For the first time in the history of the world, the

Practically, the first thing to do is to get the right kind of information. This is the first step in any investigation. The second step is to analyze the information. The third step is to draw conclusions. The fourth step is to take action. The fifth step is to evaluate the results. The sixth step is to repeat the process if necessary. This is a general outline of the investigation process. The specific steps may vary depending on the nature of the investigation.

ASSOCIATION: *Constitutive expression of the *hsp70* gene in *Escherichia coli* is controlled by the *hsp70* promoter*

RECEIVED: *John C. H. Smith, Jr.* (Signature) *John C. H. Smith, Jr.* (Signature)
TYPEWRITER: *John C. H. Smith, Jr.*

ALLIK, A., 1900-1940, Riga, Latvia, 1940.

Indicate more precisely the mechanical damage of aircraft
Mazaleivs, prop. 1940 no.1215-1c L-16. (MRA 1940)

1. Vt. dozurnyj, who may instant push bay by proxy element.

GOLIK, M. G.

"Scientific Principles Underlying the Storage of Ear and Shelled Corn and Their Practical Application." Dr Agr Sci, Khar'kov Agricultural Inst, Moscow, 1953. (RZhBiol, No 5, Nov 54)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (11)

SO: Sum. No.521, 2 Jun 55

Chemical Abst.
Vol. 48 No. 9
Apr. 25, 1954
Fats, Fatty Oils, Wax, etc.
Detergents

GOLIK M.G.

VORONTSOV, O.S.; GOLIK, M.G.; DELIDOVICH, V.N.; KLETZEV, I.A.; ZOK'-
MINA, N.P., doktor biologicheskikh nauk, professor; SOSEDOV, N.I.
FESTA, N.Ya.; CHUKHAR'KO, Z.T.; GEL'MAN, D.Ya., reizaktor; LA-
BUS, G.A., tekhnicheskij redaktor.

[Grain storage; management and equipment] Organizatsija i tekhnika
khranenija zerna. Moskva, Izd-vo tekhn. i ekonomicheskoi lit-ry.
1954. 358 p. [Microfilm] (MLRA 7:10)
(Grain--Storage)

Golik, M.G.

U C S R

✓ Accumulation of fat and its distribution in the corn grain.
M. G. Golik. Biokhim. Zerna, Akad. Nauk SSSR, 1954.
In the course of vegetat. both grain contains much fat, reaching 1.87% at ripeness. Embryo tissues contain more fat than do other parts of the grain. The embryo also bears the greater proportion of enzyme systems, which are of importance in storability of corn. Thus the state of the embryo is more important as a storability factor in corn than in other grain. G. M. K...

BAUM, A., kandidat tekhnicheskikh nauk; GOLIK, M., kandidat sel'sko-khozyaystvennykh nauk.

Transference of moisture in stored grain. Muk.-elev.prom. 20 no. 3:3-6 Mr '54.
(MERA 7:7)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut zerna i produktov ego pererabotki.
(Grain-Storage)

Gr. C. L. T. R. M. G.

Changes in the enzymic activity of ripening corn seeds.
M. G. Golik. *Trudy Vsesoyuz. Nauch. Issledovatel. Inst.*

Zerna 1954, No. 27, 66-72; *Referat. Zhur. Khim., Biol. i*

Khim. 1954, No. 11792. --The activity of dehydrogenase, α -

and β -amylase, and of lipase in the corn seeds during the

period of milk formation is high. As the seeds ripen the

activity of such enzymes gradually weakens until at the stage

of complete ripening enzymic activity is not detected. It is

suggested that the possible persistence of amylase may be

one of the factors responsible for the easy spoilage of corn

harvested before its full maturity. B. S. Leymer.

GOLIK, M. G.

Golik, M. G.: Fiziologicheskie i khimicheskie osnovy khraneniya kukuruzy (Physiological-Biochemical Bases for the Storage of Corn). Moscow: Izdatel. Akad. Nauk S.S.R., 1955. 223 pp.

GOLIK, M., doktor sel'skokhozyaystvennykh nauk.

Particular aspects of corn storage. Muk.-elev.prom. 21 no.10:4-7
0 '55. (MLRA 9:1)

1.Vysshaya zagotovitel'naya shkola.
(Corn (Maize)--Storage)

GOLIK, Mikhail Grigor'yevich, doktor sel'skokhozyaystvennykh nauk, professor;
KRETOVICH, V.L., professor, doktor biologicheskikh nauk, redaktor;
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